

## TLV of Mixtures\*

$$\frac{C_1}{T_1} + \frac{C_2}{T_2} + \dots + \frac{C_n}{T_n}$$

C = Concentration of Chemical [ppm or mg/m<sup>3</sup>]

C<sub>1</sub> = Concentration #1 [ppm or mg/m<sup>3</sup>]

T<sub>1</sub> = TLV #1 [ppm or mg/m<sup>3</sup>]

When a worker is exposed to more than one chemical and each chemical effects the same target organ [i.e., central nervous system] then the sum of their ratios must be less than one [1] or the combined exposures are considered to be exceeding the TLV of the mixture.

It is a serious mistake to compare each chemical in a mixture independently to their respective TLVs. To do so dramatically underestimates the actual risk to the exposed workers.

### Example [ppm]:

- C<sub>1</sub>=14, T<sub>1</sub>=25; C<sub>2</sub>=1.8, T<sub>2</sub>=5; C<sub>3</sub>=1, T<sub>3</sub>=10  
TLV=14/25 + 1.8/5 + 3/10  
TLV=1.22; TLV of mixture is exceeded.
- C<sub>1</sub>=5, T<sub>1</sub>=25; C<sub>2</sub>=3, T<sub>2</sub>=5; C<sub>3</sub>=1, T<sub>3</sub>=10  
TLV=5/25 + 3/5 + 1/10  
TLV=0.9; TLV of mixture is not exceeded.

\*ACGIH®, TLVs, <https://www.acgih.org/>